

# 603 Chemistry Calorimetry Lab Answers

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October 19th, 2014 - 2 535 x 1 mol 40 g 0 063375 moles NaOH 4 Yes because the enthalpy change should double my prediction 5 NaOH would need a high amount of energy to break apart because its an ionic bond Ionic bonds are stronger than covalt bonds 6 Since there s a gap that mean heat would be escaping from

## CALORIMETRY AND SPECIFIC HEAT LAB ANSWERS REPORT

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March 22nd, 2015 - 6 03 Calorimetry No description by gabriela cardenas on 23 March 2015 Using the specific heat capacity value that I determined in Part II of the lab I believe the most probable identity of the unknown metal is tin with a heat capacity of 0 210 Tin s heat capacity of 0 210 is the number closest to the specific heat capacity of the unknown metal of 0 222

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### **Calorimetry Chemistry for Majors Lumen Learning**

November 29th, 2018 - One technique we can use to measure the amount of heat involved in a chemical or physical process is known as calorimetry Calorimetry is used to measure amounts of heat transferred to or from a substance

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