

# 43 How Atoms Differ Answers

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December 2nd, 2018 - Different atoms either have a different number of protons which makes them different elements or a different number of neutrons which makes them isotopes of the same element For example hydrogen has one proton and usually no neutrons while helium has two protons and usually two neutrons

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about atomic number For each statement below write true or false 1 The number of neutrons in an atom is referred to as its atomic number 2 The periodic table is arranged by increasing atomic number 3 Atomic number is equal to the number of electrons in an atom 4 The number of protons in an atom identifies it

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#### **How do an atom of hydrogen and a molecule of hydrogen differ**

November 26th, 2018 - Best Answer An atom of hydrogen is indicated by the symbol H A molecule is indicated by H<sub>2</sub> The molecule has two atoms bonded to each other for stability Where one H atom has one electron and an empty space in its electron shell an H<sub>2</sub> molecule has two electrons shared between them so that they both have a filled electron shell

#### **Describe how an atom differs from a molecule Yahoo Answers**

December 3rd, 2018 - Best Answer Atoms are the building blocks of molecules A molecule is essentially multiple atoms joined together An atom can be defined as the smallest part of an element or a compound that can take part in a chemical reaction A molecule can be defined as the smallest part of an element or a compound that can exist in a free

#### **How do Atoms Differ from one another Yahoo Answers**

November 15th, 2018 - Best Answer Crap Face is correct along with the shape of the orbitals are the energy states of the electrons within the atoms But there is something more that can be said that is in a way quite interesting In the current way of seeing subatomic particles if you have two atoms of the same element

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s t i h l w a t e r b l a s t e r m a n u a l  
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